Temperature Distribution in Copper Electrode during Electrical Discharge Machining Process

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Abstract: This paper reports on a new method used to estimate the spark radius in the gap during electrical discharge machining (EDM). This method combines the heat flux and energy equations of the copper electrode. The energy partition between workpiece and electrode (tool) due to EDM process was estimated using the ratio of thermal conductivity of the workpiece to that of the electrode. Using the energy partition between the electrode and the workpiece, temperature distribution in the electrode was established and Gaussian heat distribution was used to analyze the energy released from a single spark. The energy released due to a single spark was used to calculate the fraction of energy received by the electrode based on its thermal conductivity. The 3-D temperature distribution in the electrode was carried out using ANSYS version 11.0 and the estimated temperature of the electrode from a single spark was validated by thermal diffusivity of the electrode material. The difference of 6% was recorded between the simulated and calculated temperatures of the copper electrode. Based on the achieved percentage error, the simulated temperature on the copper electrode can be accepted as EDM process.

Keywords: EDM process; Heat flux; Spark radius; Temperature distribution; Thermal conductivity

1. Introduction

Electrical discharge machining (EDM) is used in precise machining of both complex and non-complex objects, as an alternative to traditional machining techniques [1]. EDM process involves combination of several facets thermodynamic, electrodynamics, such as hydrodynamic and electromagnetic these make EDM process difficult to present in a simple way [2]. EDM is used in precise machining of both complex and non-complex objects (such as tool and dies, injector nozzles and turbine blades in aircraft) as an alternative to traditional machining techniques [1]. EDM process involves combination of several facets such as thermodynamic, electrodynamics, hydrodynamic and electromagnetic these make EDM process difficult to present in a simple way [2].

According to Yeo et al., [3] many works on the analytical modeling of the EDM process have been carried out by many researchers [4-13] through the use of electro-thermal and electro-mechanical approaches. The energy generated by electrical discharge is shared between the cathode, anode, and the plasma column. Different heat input methods have been adopted by various researchers to predict the temperature distribution in both electrodes [5, 8, 12-16].

This study analyzed the temperature distribution in copper electrode during EDM with and without liquid nitrogen. The new thing in this paper is that the analysis was carried out by combining the heat flux equation and energy received equation for copper electrode to estimate EDM spark radius. The estimated temperature of the electrode was validated by thermal diffusivity of the electrode material. The validated result shows good agreement between the simulated and measured temperature of the electrode.

2. Heat Input

In this study, Gaussian heat input model (Fig. 1) has been used to estimate the heat from the plasma. The model gives the best approximation of EDM plasma. For Gaussian heat distribution, heat flux is given by Eq. 1 [5, 7-9, 11, 12].

Heat flux
$$q_f(R) = \frac{4.45 W_M IV}{\pi (r_{sp})^2} \times e^{\left[-4.5\left(\frac{R}{r_{sp}}\right)^2\right]}$$
 (1)

Where q_f = heat flux (W/mm²); W_M = fraction of energy utilized by the material (Watt); I = pulse current (Amp); V = gap voltage (Volt); R= radial distance from the axis of the spark (µm); r_{sp} = spark radius radius (µm).



Fig. 1. Gaussian heat distribution for modeling the electrode.

3. Energy Release from a Single Spark

Energy E_{sp} (watt) released from a single spark is given by Eq. 2 [4, 5].

$$E_{sp} = I \times V \times t_{on} \tag{2}$$

Where I = pulse current (Amp); V = gap voltage (Volt); $t_{on} =$ on-time (µs). According to [4, 5], the total energy absorbed by any material from single spark is given by Eq. 3.

$$E_M = W_M \times I \times V \times t_{on} \tag{3}$$

 E_M = total energy receive by the material (Watt); W_M = fraction of energy absorbed by the material (Watt). In this work, spark radius (r_{sp}) was calculated by combining Eq. 1 (heat flux equation) and Eq. 3 (energy absorb equation) based on the fact that plasma radius is a function of current intensity, discharge voltage, pulse duration and fraction of energy absorbed by the material [4, 5, 7, 12].

4. Energy Partition

Titanium workpiece absorbs less heat due to its lower thermal diffusivity compared to the copper electrode. In EDM modeling, many researchers used different fractions of heat distribution; ranges from 17% to 20% [4, 5, 12, and 14]. The ratio of thermal conductivity of the workpiece to that of electrode was used to obtain the percentage of energy fraction to be absorbed by the electrode.

 $\frac{Thermal \ conductivity \ of \ workpiece}{Thermal \ conductivity \ of \ electrode} = \frac{21.59}{401.19} = \frac{1}{19} = 1:19$

Fraction of energy absorbed by the electrode is taken to be 0.19. This is in line with the work of [4].

5. Heat Flux

Spark radius r_{sp} is estimated by combining Eq. 3 and Eq. 1. The values of 5.3A, 4.3µs and 23V for the current, pulse on-time and voltage respectively were the values used for EDM machining. These values were selected based on the available variations in machine settings.

Energy received by copper (Eq. 3) = Heat flux into copper (Eq. 1)

$$W_{cu} \times I \times V \times t_{on} = \frac{4.45 W_{cu} IV}{\pi (r_{sp})^2} \times e^{\left[-4.5 \left(\frac{R}{r_{sp}}\right)^2\right]}$$
(4)

 W_{cu} is fraction of energy that goes into copper, *I* is current (A), *V* is gap voltage (V), t_{on} is pulse on-time (µs), *R* (Fig. 1) is radial distance of the point under consideration from the axis of spark (mm) and r_{sp} is spark radius (µm). Just before heat flux into the electrode, the value of *R* is zero; hence Eq. 4 reduced to:

$$W_{cu} \times I \times V \times t_{on} = \frac{4.45 W_{cu} I V}{\pi (r_{sp})^2}$$

Therefore, spark radius $r_{sp} = \sqrt{\frac{4.45}{\pi \times t_{on}}} = 0.574 \ \mu m$

Since a good estimate of radial distance R from the axis of the spark cannot be made, the following assumptions were made:

(1) The radial distance R from the axis of the spark is taken to be equal to spark radius of 0.0005 mm.

(2) The spark occurs within a minimum gap size of 0.1 mm [17, 18].

To ensure that heat is concentrated at a point within the gap.

From Eq.1 Heat flux
$$q_{f}(R) = \frac{4.45 W_{cu} IV}{\pi (r_{sp})^{2}} \times e^{\left[-4.5 \left(\frac{R}{r_{sp}}\right)^{2}\right]}$$

= $\frac{103.06645}{1.035079381 \times 10^{-6}} \times e^{\left[-5.445\right]} = 429,942.33 \text{ W/mm}^{2}$

The estimated heat flux into the copper electrode is 429.9 KW/mm^2 .

6. Temperature Modeling

Due to random and complex nature of EDM, the following assumptions were made: (i). the analysis is made for one spark, (ii). Total power of each pulse is used by one spark, (iii) Electrode domain is axisymmetric, (iv). Heat transfer to electrode is by conduction, (v). There is constant heat flux into the electrode, (vi). The temperature distribution is governed by heat conduction. Conduction heat transfer serves as the thermal boundary condition on surface 1 (Fig. 1). No heat transfer across boundaries 2, 3 and 4.

<u>Initial condition</u>: At time t = 0; Temperature $T = T_{28}{}^{0}{}_{C}$ (ambient temperature) at electrode domain. <u>Boundary conditions</u>: At time t > 0; on boundary 1, there exist:

nitrogen was introduced. The exit temperature was observed for the two conditions during EDM. A rapid change in temperature is observed within a very small area in the vicinity of the applied temperature for both electrodes. As the distance x is farther away from the input temperature, there is a corresponding change in the values of the temperature along the electrode length up to the point where a temperature of -101° C and 28° C were recorded for both electrodes (with and without nitrogen) respectively.



Fig. 2 Electrode heat flux with boundary conditions.

8. Temperature Validation

Distance i-j = 5mm;

The temperature of $1,072^{\circ}C$ was validated by heat equation. From Fig. 2: Point i is at temperature T₁; Point j is at temperature T₂ = $430^{\circ}C$ (T₁ > T₂)

 $K \frac{\partial T}{\partial z} = \begin{cases} q_{R} \text{ if } R \leq r_{R} \text{ pulse-off time = 0} & \frac{\partial T}{\partial n} = 0 \text{ on boundaries } 2, 3 \& 4 \\ KA \frac{\Delta T}{\Delta z} \text{ if } R > r_{R} \text{ (dist. betw electrode \& spark is assumed small)} m^{2}/\text{(s)} \\ \Delta Z = \text{distance (mm)} \end{cases}$

From
$$q_{conv} = KA \frac{\Delta T}{\Delta z}$$
 (A & Δz are negligible) (5)
$$T = \frac{429,942.33}{401} = 1,072.18^{\circ}C$$

Temperature at point *i* on the copper electrode is estimated to be $1,072.18^{\circ}$ C.

7. Results

Figures 3 and 4 show the temperature variations along copper electrode when liquid

Thermal diffusivity is given by
$$\alpha = \frac{k}{\rho C_p}$$
 (7)

k = thermal conductivity (W/mm) ρ = density of copper (g/mm³) C_p = specific heat capacity (J/g ⁰C)

$$\alpha = \frac{401}{1000} \times \frac{1000}{8.96} \times \frac{1}{0.385} = 116.25 \ mm^2 \ / \ s$$

From (Eq. 6) $T_1 = 403 + 116.25 (5) = 1,011.3^{\circ}C$



Fig. 3 Variations in temperature distributions on the electrode: (a) EDM with nitrogen, (b) EDM without nitrogen.

9. Conclusion

The analytical results of the analysis of temperature distribution in copper electrode during EDM process have been presented. The main conclusion is that, the temperature dependency of the electrode is very important to the accuracy of the results and this gives better correlations with the simulated temperature. Furthermore, the estimated temperature of 1072^{0} C for the electrode was validated giving 6% error between the two temperatures. Since there is no much disagreement between the simulated value of 1072^{0} C and experimental value of $1,011.3^{0}$ C, the temperature can be accepted to close to EDM process.

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